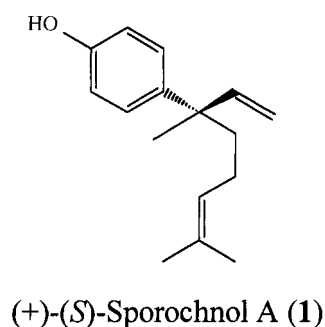


A facile synthesis of (\pm)-sporochinol A[†]Ying Li^{a*}, Hao Yuan^a, Biao Lu, Yulin Li^a and Dawei Teng^b^aNational Laboratory of Applied Organic Chemistry and Institute of Organic Chemistry, Lanzhou University, Lanzhou 730000, P.R. China^bDepartment of Chemistry, Qindao University, Qindao 266071, P.R. China

A facile synthesis of (\pm)-sporochinol A, starting from readily available monoketal of 1,4-cyclohexanedione in high overall yield, is described. The key step is the alkylation of monoketal of 1,4-cyclohexanedione with geranyl bromide mediated by chromium(II).

(+)-Sporochinol A (**1**), first isolated by Fenical and co-workers in 1993 from the Caribbean marine alga *Sporochnus bolleanus*,¹ has been shown to exhibit significant feeding deterrence towards herbivorous fish.



(+)-Sporochinol A (**1**) has a novel structure containing a prenylated phenol and an asymmetric quaternary carbon. The chemical structure and absolute stereochemistry were determined by spectral and chemical methods. Its absolute configuration assigned to be (*S*) by synthetic method.² Recently enantioselective synthesis of **1** by Fadel^{2,3} and Ogasawara⁴ has been investigated. Herein we report a facile synthesis of (\pm)-**1** from the monoketal of 1,4-dicyclohexanedione and geranyl bromide.

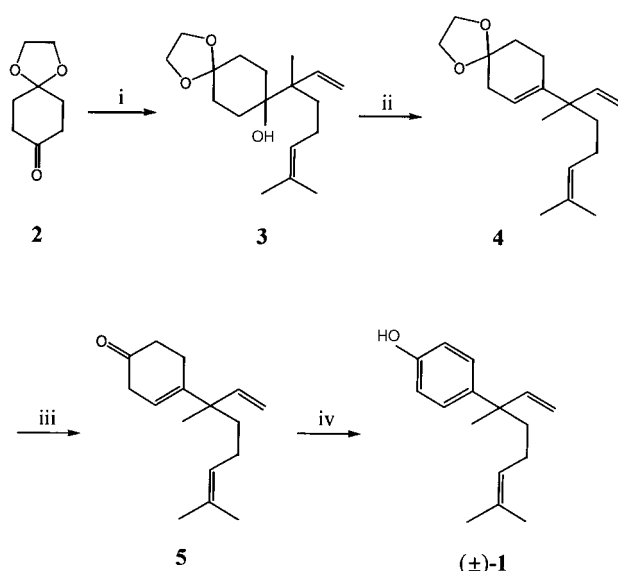
The synthetic route of (\pm)-**1** is outlined in Scheme 1.

Reagents and conditions: (i) 1) CrCl₃, LiAlH₄, THF, 0°C, 2) geranyl bromide, DMF, r.t. 6h, 74%; (ii) SOCl₂, pyridine, benzene, 0°C, 10min, 92%; (iii) *p*-TsOH (cat.), acetone, reflux, 3h, 90%; (iv) PdCl₂, Na₂CO₃, *tert*-butanol, 70°C, 10h, 79%.

1,4-Cyclohexanedione monoketal **2**,⁵ readily obtained by selective ketalization of 1,4-cyclohexanedione, was alkylated with geranyl bromide mediated by chromium(II)⁶ in DMF to give the alcohol **3** in 74% yield.⁷ The alcohol **3** was converted into the corresponding alkene **4** by dehydration with SOCl₂ in pyridine.⁸ Deprotection of ketal **4** by normal acidic hydrolysis to give ketone **5**, which was exposed⁹ to PdCl₂ in *tert*-butanol containing 4 equivalent of Na₂CO₃ at 70°C to afford (\pm)-sporochinol A (**1**) in 79% yield, and in 48% overall yield from monoketal **2**. The synthesis described here is short, and efficient.

Experimental

¹H NMR spectra were recorded on a Varian FT-80A or a Bruker AM-400 spectrometer in CDCl₃ solution using TMS as the internal reference. IR spectra were obtained on a Nicolet AVATAR 360 FT-IR (film) spectrometer. Mass spectra were measured on a VG ZAB-HS spectrometer by direct inlet at 70 eV. Melting points were measured on a Kofler hot stage and uncorrected. All solvents were freshly



Scheme 1

purified and dried by standard techniques prior to use. All reactions were routinely carried out under an inert atmosphere of Ar, and monitored by TLC. Purification of products were conducted by flash column chromatography on silica gel (200–300 mesh) purchased from Yan Tai Yuan Bo Silica Gel Co. or Al₂O₃ purchased from Shang Hai Wu Si Chemical Reagents Company, China.

8-(1-Ethyl-1,5-dimethyl-4-hexenyl)-1,4-dioxaspiro[4,5]dec-8-ol (**3**): Lithium aluminum hydride (0.49 g, 12.8 mmol) was added portionwise to a suspension of chromium(III) chloride (4.0 g, 25.6 mmol) in THF (30 ml) at 0°C over 15 minutes. After the gas evolution had ceased, the reaction mixture was concentrated under reduced pressure, and the resulting residue was taken up in DMF (20 ml). To the above chromium(II) reagent was added the monoketal **2** (1.0 g, 6.4 mmol) and then a DMF (10 ml) solution of geranyl bromide (2.78 g, 12.8 mmol) at room temperature. After 3h the reaction mixture was treated with brine (30 ml) and extracted thoroughly with ether (3×100 ml). The organic phases were dried over anhydrous MgSO₄ and concentrated. The crude product was chromatographed on Al₂O₃ eluting with pet. ether:ethyl ether (v/v 15:1) to afford alcohol **3** (1.39 g, 74%) as white solid. m.p. 64–65°C; IR (film): ν_{\max} /cm⁻¹ 3499, 3079, 2966, 1634, 1375, 1101, 1037, 889; ¹H NMR (CDCl₃, 400MHz) δ (ppm): 5.9 (dd, $J_{\text{trans}} = 17.2$ Hz, $J_{\text{cis}} = 10.8$ Hz, 1H), 5.31 (d, $J_{\text{cis}} = 10.8$ Hz, 1H), 5.15 (d, $J_{\text{trans}} = 17.2$ Hz, 1H), 5.09 (t, 3H), 3.95 (s, 4H), 2.0–1.4 (m, 12H), 1.67 (br s, 3H), 1.58 (br s, 3H), 1.06 (s, 3H); EIMS (m/z): 294 (M⁺, 0.1), 276 (M⁺-18, 0.4), 194 (6.6), 169 (9), 157 (31), 138 (17), 123 (25), 100 (20), 95 (100), 69 (48).

8-(1-Ethyl-1,5-dimethyl-4-hexenyl)-1,4-dioxaspiro[4,5]dec-7-ene (**4**): Thionyl chloride (0.41 ml, 5.65 mme) was added dropwise to a stirred solution of alcohol **3** (1.37 g, 4.65 mmol) in anhydrous benzene (25 ml) and pyridine (5 ml) at 0°C and the reaction mixture was stirred at 0°C for 10 minutes. It was then treated with ice water (20 ml) and extracted thoroughly with ether (3×50 ml). The organic phases were washed with 5% HCl (3×10 ml), saturated NaHCO₃ (10 ml), H₂O and brine, and dried. Evaporation of the solvent gave the

* To receive any correspondence. E-mail: liyl@lzu.edu.cn

[†] This is a Short Paper, there is therefore no corresponding material in *J. Chem. Research (M)*.

residue, which was chromatographed on Al_2O_3 eluting with pet. ether:ethyl acetate (v/v 30:1) to afford **4** (1.18 g, 92%) as a colourless oil. IR (film): $\nu_{\text{max}}/\text{cm}^{-1}$ 3080, 3053, 2966, 1633, 1375, 1258, 1116; ^1H NMR (CDCl_3 , 80MHz) δ (ppm): 5.9 (dd, $J_{\text{trans}} = 17.2$ Hz, $J_{\text{cis}} = 10.8$ Hz, 1H), 5.42 (t, 1H), 5.10 (t, 1H), 5.06 (d, $J_{\text{cis}} = 10.8$ Hz, 1H), 4.9 (d, $J_{\text{trans}} = 17.2$ Hz), 3.98 (s, 4H), 2.3–1.4 (m, 10H), 1.7 (br s, 3H), 1.6 (br s, 3H), 1.13 (s, 3H); EIMS (m/z): 276 (M^+ , 4), 233 (4), 194 (54), 149 (8), 131 (13), 105 (2), 99 (57), 86 (100), 55 (22), 41 (18). HRMS calcd for $\text{C}_{18}\text{H}_{28}\text{O}_2$: 276.2090; Found: 276.2066.

4-(1-Ethyl-1,5-dimethyl-4-hexenyl)-3-cyclohexen-1-one (**5**): A mixture of ketal **4** (1.16 g, 4.22 mmol) and a catalytic amount *p*-TsOH in dry acetone (30 ml) was refluxed for 3h. The mixture was evaporated *in vacuo* and the residue was chromatographed on silica gel eluting with pet. ether:ethyl acetate (v/v 30:1) to give the ketone **5** (0.88 g, 90%) as a colourless oil. IR (film): $\nu_{\text{max}}/\text{cm}^{-1}$ 3080, 2968, 2936, 1720, 1633, 1193; ^1H NMR (CDCl_3 , 400MHz) δ (ppm): 5.87 (dd, $J_{\text{trans}} = 17.2$ Hz, $J_{\text{cis}} = 10.8$ Hz, 1H), 5.59 (t, 1H), 5.11 (t, 1H), 5.06 (d, $J_{\text{cis}} = 10.8$ Hz, 1H), 4.78 (d, $J_{\text{trans}} = 17.2$ Hz, 1H), 2.93 (d, 2H), 2.42 (br s, 4H), 2.0–1.6 (m, 4H), 1.71 (s, 3H), 1.61 (s, 3H), 1.18 (s, 3H); EIMS (m/z): 232 (M^+ , 4), 189 (12), 161 (5), 150 (41), 121 (14), 105 (35), 93 (44), 91 (43), 83 (58), 55 (100), 41 (77). HRMS calcd for $\text{C}_{16}\text{H}_{24}\text{O}$: 232.1827; Found: 232.1857.

(\pm)-*Sporochinol A* (**1**). A mixture of ketone (**5**) (200 mg, 0.86 mmol), PdCl_2 (150 mg, 0.86 mmol), and anhydrous Na_2CO_3 (360 mg, 3.44 mmol) in *tert*-butanol (20 ml) was refluxed for 8h. The reaction mixture was quenched with water (10 ml) and extracted with ether (3 \times 50 ml). The organic phases were washed with H_2O , brine, and dried. Evaporation of the solvent gave the residue, which was chromatographed on silica g. Elution with pet. ether:ethyl acetate (v/v 20:1) afforded (\pm)-**1** (150 mg, 79%) as a colourless oil. IR (film): $\nu_{\text{max}}/\text{cm}^{-1}$ 3386, 2924, 1635, 1608, 1594, 1512, 1440, 1242, 1178, 913, 830; ^1H NMR (CDCl_3 , 400MHz): δ (ppm) 7.18 (d, $J = 6.6$ Hz, 2H), 6.76 (d, $J = 6.6$ Hz, 2H), 6.0 (dd, $J_{\text{trans}} = 17.7$ Hz, $J_{\text{cis}} = 10.7$ Hz,

1H), 5.08 (t, 1H), 5.06 (d, $J_{\text{cis}} = 10.7$, 1H), 5.02 (d, $J_{\text{trans}} = 17.7$ Hz, 1H), 4.75 (br s, 1H, OH), 1.8–1.6 (m, 4H), 1.67 (br s, 3H), 1.51 (br s, 3H), 1.34 (s, 3H); EIMS (m/z): 231 (M^+ +1,1), 230 (M^+ , 6), 187 (4), 148 (22), 147 (100), 120 (13), 107 (17), 91 (14), 83 (12), 41 (24). HRMS calcd for $\text{C}_{16}\text{H}_{22}\text{O}$: 230.1664; Found: 230.1681.

This research was financially supported by Foundation for University Key Teacher by the Ministry of Education.

Received 21 October 2000; accepted 15 November 2000
Paper 00/553

References and notes

- 1 Y.C. Shen, P.I. Tsai, W. Fenical and M.E. Hay, *Phytochemistry*, 1993, **52**, 71–75.
- 2 M. Takahashi, Y. Shioura, T. Murakami and K. Ogasawara, *Tetrahedron: Asymmetry*, 1997, **8**, 1235–1242.
- 3 T. Kamikubo, M. Shimizu and K. Ogasawara, *Enantiomer*, 1998, **2**, 297–301.
- 4 A. Fadel and L. Vandromme, *Tetrahedron: Asymmetry*, 1999, **10**, 1153–1162.
- 5 P. Mussini, F. Orsini and F. Pedizzoni, *Synth. Commun.*, 1975, **5**(4), 283–286.
- 6 T. Hiyama, Y. Okude, K. Kimura and H. Nozaki, *Bull. Chem. Soc. Jpn.*, 1982, **55**, 561–568.
- 7 The Grignard addition of geranyl magnesium bromide to monoketal **2** gave the desired product **3** in low yield (<20%).
- 8 A. Terada, Y. Tanoue, A. Hatada and H. Sakamoto, *Bull. Chem. Soc. Jpn.*, 1987, **60**, 205–213.
- 9 B. Bierling, K. Kirschke and H. Oberender, M. Schulz, *J. Prakt. Chem.*, 1972, **314**(1), 170–180.